Alcohol, Phenol, Ether Problems

On the following compound identify the primary, secondary, tertiary, and quaternary carbons.				
CH ₃ CH ₂ CH ₂ CH(C	H ₃)CH ₃			
	olecules. Label each as ei	ther an alcohol, a phenol, o		
an ether.				
CH ₃ -O-CH ₃	CH ₃ CH ₂ -OH			
CH₃OH		CH ₃ CH ₂ -O-CH ₃		
	CH ₃ CH ₂ CH ₂ CH(C Shown below are several m an ether. CH ₃ -O-CH ₃	CH ₃ CH ₂ CH ₂ CH(CH ₃)CH ₃ Shown below are several molecules. Label each as ei an ether. CH ₃ -O-CH ₃ CH ₃ CH ₂ -OH		

 CH_3

OH

Shown below are the structures of several alcohols. Tell whether the alcohol is 4. primary, secondary, or tertiary.

CH₃OH CH₃ - CH

 CH_3 CH_3 - C - CH_3 CH_3CH_2OH OH

5. Write the IUPAC names of the following alcohols.

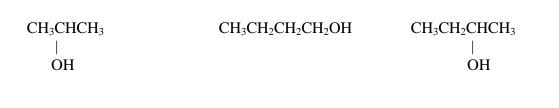
> CH₃OH CH₃CH₂OH CH₃CH₂CH₂OH

CH₃CHCH₃ CH₃CH₂CH₂CH₂OH CH₃CH₂CHCH₃ OHOH

How does hydrogen bonding explain the elevated boiling point of alcohols? 6.

7	Label each	compound a	s soluble	slightly	soluble	or insolub	de in	water
1.	Lauci cacii	compound a	s soluble,	Singing	Soluble	or msorue	111 Ju	water.

CH₃OH CH₃CH₂OH CH₃CH₂CH₂CH₂CH₂OH



8. Label each of the following compounds as neutral or acidic.

CH₃-O-CH₃ CH₃CH₂-OH

9. For each of the following compounds tell how long it would take for a white cloudy layer to form when mixed with the Lucas Reagent.

CH₃OH

CH₃ - CH | OH

 CH_3



10. Predict the organic product for the following reactions.

$$CH_3$$
|
 $CH_3 - C - CH_3 + Cr_2O_7^{-2} \Rightarrow$
|
 OH

$$CH_3CH_2OH + Cr_2O_7^{-2} \rightarrow$$

$$\begin{array}{c} CH_3 \\ | \\ CH_3 - CH \\ | \\ OH \end{array} + Cr_2O_7^{-2} ~ \rag{9}$$

- 11. What color change does one observe when acidified dichromate reacts with an alcohol?
- 12. Which of the following compounds will react with acidified dichromate solution?

CH₃OH

CH₃OH

CH₃ |

CH



OH 	OH
ОН 	OH
Write out the equati	on for the reaction of phenol with water.

6.	Label each of the following ethers as either aliphatic or aromatic.					
	CH ₃ CH ₂ -O-	-O-	CH ₃ -O-CH ₂ CH ₃			
7.	Name each of the following ethers.					
	CH ₃ CH ₂ -O-	-O-	CH ₃ -O-CH ₂ CH ₃			
	CH ₃ -O-CH ₃	CH ₃ CH ₂ -O-CH ₂ CH ₃	CH₃-O-			